

Chapter 19:-

19.1:-

The two observers in relative motion will always agree upon:

The speed of light $c = 3 \times 10^8 \text{ ms}^{-1}$
The force acting
The laws of Physics.

19.2:-

Yes, time really passes more slowly in moving system due to relative speed. The proper time (t_0) and time (t) are related as:

$$t = \frac{t_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

$$\text{As } \sqrt{1 - \frac{v^2}{c^2}} < 1$$

$$\text{So, } t > t_0$$

19.3: ^{v < c} imp

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i) If we are moving in a spaceship at a very high speed relative to earth. There will not be any difference in our pulse rate because the ^{frame} phase of reference is same as the spaceship.

ii) There will be difference in the pulse rate of people on earth because the frame of reference are different.

19.4:

According to the

special theory of relativity

$$t = \frac{t_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

if $c \rightarrow \infty$
in S_0 ,

$$t = \frac{t_0}{\sqrt{1-0}} \Rightarrow t = t_0$$

And,

$$l = \frac{l_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

if $c \rightarrow \infty$
in S_0 ,

$$l = l_0 \sqrt{1-0}$$

$$l = l_0$$

And,

$$m = \frac{m_0}{\sqrt{1 - \frac{v^2}{c^2}}}$$

if $c \rightarrow \infty$
in S_0 ,

$$m = \frac{m_0}{\sqrt{1-0}} \Rightarrow m = m_0$$

And,

$$E = mc^2$$

if $c \rightarrow \infty$

Then $E = \infty$

19.5:-

Yes, we can conclude that a compressed spring has more mass than the ^{same} spring when it is not compressed.

Because compressed spring has elastic P.E that slightly increases its mass.

19.6:-

As the solid is heated and begins to glow, it emits radiations having energy according to temperature provided. As the temperature increases the wavelength of emitted radiations will decrease.

So, firstly radiations of greater wavelength will

emitt that are red
in colour.

19.7:-
According to Stefan
Boltzmann law

$$E = \sigma T^4$$

is absolute temperature
is doubled.

$$E' = \sigma (2T)^4$$

$$= 16\sigma T^4$$

$$E' = 16E \quad \because E = \sigma T^4$$

So the total radiation
energy will be increased
by 16 times when absolute
temperature is doubled.

19.8:-
According to Planck's

$$E = hf$$

$$E = \frac{hc}{\lambda}$$

$$E_n = n \frac{hc}{\lambda}$$

$$n = \frac{\lambda E_n}{hc}$$

$$n \propto \lambda$$

As the wavelength for red light is greater than blue light, it will have greater number of photons.

$$\lambda_{red} > \lambda_{blue}$$

So,

$$n_{red} > n_{blue}$$

19.9:-

We know that:

$$E = hf \quad \text{and} \quad E = pc$$

As the blue color has greater frequency than red or green. So, photon of blue carries the most energy and momentum.

19.10:-

As the frequency

of radio waves is less
than frequency of x-rays

So, radio waves have
lower energy quanta

$$E = hf$$

f
radio
waves

$<$
 f
x-rays

$$E < E$$

radio
waves

x-rays

19.11:-

The brightness of a
beam of light primarily
depends upon the number
of photons

Because the number
of photons determines the
intensity of light that
determines the brightness.

19.12:-

Infrared light has energy
less than visible

light.

So, when infrared light falls on the Dyes, visible light can not be emitted.

Whereas, ultraviolet light has energy greater than visible. So, they can be visible light.

19.13:-

Yes, bright light will eject more electrons from a metal surface than dimmer light of the same colour.

Because the bright light will contain more photons to eject more number of electrons.

number of
Brightness \propto photons \propto no. of emit electrons.

19.14:-

No, Higher frequency light will not eject greater no. of electrons than low frequency light. Number of ejected electrons will be same in both case. Frequency determines the energy of ejected electrons.

Frequency \propto energy of ejected electron.

19.15:-

Yes, momentum is transferred in the metal surface. When light shines on the surface.

Because when light shines on metal surface photons falls on the metal and transfer energy and momentum.

mathematically:-

$$E = hf \quad \text{and} \quad E = Pc$$

19.16:-

Red light has smaller energy than blue or white

light due to smaller frequency. So red light will not disturb the developing films in the photographic dark room.

Therefore, the ^{red} light can be used in dark room but blue or white light contain greater energy and can not be used.

19.17:-

Given that:

$$E_a = 2E_b$$

we know that:

$$P_a = \frac{E_a}{c} \quad \text{and} \quad P_b = \frac{E_b}{c}$$

$$P_a = \frac{2E_b}{c}$$

$$= 2 \left(\frac{E_0}{c} \right)$$

$$P_A = 2 P_B$$

So the momentum of photon A will also be doubled than photon B.

$$\frac{P_A}{P_B} = 2$$



$$P_A : P_B = 2 : 1$$

19.18:-

Compton effect can not be observed with visible light because it has less energy and momentum to emit electrons and Compton wavelength to observe the Compton effect X-rays are used.

19.19:-

No, Pair production can not be take place in vacuum.

Because for the pair

production There should be
a nucleus which breaks
down into electron and positron
due to striking of γ -rays.
But vacuum there will
not be any nucleus. So pair
production can not be
observed.

19.20:-



— No, it is not possible
to create a single electron
from energy.

Because it is
against the law of conservation
of energy. When nucleus breaks
down it will produce
electrons with anti-~~positron~~ electrons
that is positron. Single
electron can not be
produced.

19.21:-

— If electrons behaves
only like particles then

They will just strike to
the screen and no
diffraction pattern will observe.

19.22:-

We know that

$$P = mv \quad \text{and} \quad P = \frac{h}{\lambda}$$

comparing both eq:-

$$mv = \frac{h}{\lambda}$$

$$v = \frac{h}{m\lambda}$$

For same wavelength :-

$$v \propto \frac{1}{m}$$

As electron has less mass
 $9.1 \times 10^{-31} \text{ kg}$ than mass of proton
 $1.67 \times 10^{-27} \text{ kg}$

Therefore, electron has
greater speed than the
proton.

19.23:-

We do not notice the de Broglie wavelength for a pitched cricket ball due its large mass.

$$\lambda = \frac{h}{mv}$$

Due to large mass; the wavelength will be small that λ can not be noticed.

19.24:-

Alpha particle has greater mass due to shorter wavelength than electron, neutron and proton.

Mathematically,

$$\lambda = \frac{h}{mv}$$

19.25:-

Light behaves as a wave for the following phenomenon.

- Reflection
- Refraction

- Diffraction
- Interference
- Polarization

light behaves as a particle for the following phenomena:

- Photoelectric effect
- Compton effect
- Pair production

19.26:

Electron microscope has following advantages over an optical microscope:

Electron microscope has great resolution of 0.5 to 1 nm as compared to optical microscope having 0.2 to 0.2 μ m.

Electron microscope has great magnification than optical microscope.

Electron microscope can produce three dimensional.

Electron microscope can produce the internal structure of object.

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19.27:

No. The measurements can not show precise momentum if they show precise position for an electron.

Because according to Heisenberg's uncertainty principle the momentum and position of an electron can not be measured accurately at same time.

$$\Delta x \cdot \Delta p \approx h$$