

NEW MODEL QUESTION PAPER

S.S.C. (ANNUAL) EXAMINATIONS

CHEMISTRY (THEORY) PAPER-I

CLASS IX (SCIENCE GROUP)

Time: 3Hours (Marks: 60)

SECTION "A" (20%) MULTIPLE CHOICE QUESTIONS (MCQs) (12 Marks)

Q.1. 12 MCQs will be given from whole syllabus. Each carries 1 Mark.

SECTION "B" (40%) (SHORT ANSWERS QUESTIONS) (24 marks)

Note: Answer any Eight (08) questions from this section. Each carries 03 marks.

- What are Isotopes? Identify the number of Protons and Neutrons present in the following Isotopes.

 37—-
- 3- What is an Alloy? Write down the composition and one use of Brass and Bronze.
- 4- Balance the following chemical equations by Inspection method.

$$KI + Cl_2 \longrightarrow KCl + I_2$$
 $CaCO_3 + HCl \longrightarrow CaCl_2 + H_2O + CO_2$
 $NaHCO_3 \longrightarrow Na_2CO_3 + H_2O + CO_2$

- 5- Design Primary & confirmative tests for qualitative analysis of Zn⁺² (Cation).
- 6- Write down electronic configuration of the following elements.
 - (i) Nitrogen
- (ii) Magnesium
- (iii) Chlorine
- 7- A 600ml sample of gas is heated from 27°C to 77°C at constant pressure. What is its final volume?
- 8- Indicate one use of the following compounds with their chemical formulae.

Slaked Lime, Lime Stone, Soda ash

- 9- State Faraday's first and second laws of Electrolysis.
- 10- What are electrochemical reactions? Describe its types on the basis of transfer of electron.
- 11- Write down three characteristics of Ionic compounds.
- 12- Calculate the number of moles and number of molecules present in 20 grams of NaOH.
- **13-** Justify:
 - i) Milk is a colloidal solution.
 - ii) Gases diffuse more rapidly.

<u>SECTION- "C"</u> (40%) (DESCRIPTIVE ANSWER QUESTIONS) (24 Marks)

Note: Answer any Four (04) questions from this section. Each carries <u>06</u> marks.

- 14- State Boyle's Law and derive its mathematical expression $P_1V_1 = P_2V_2$.
- 15- Define Covalent bond and on the basis of number of bond pair classify it into three types with examples.
- **16-** Discuss in detail the long form of periodic table.
- 17- What are Crystalline solids and Amorphous solids and Write any four differences between: Crystalline Solids & Amorphous Solids.
- 18- What is Corrosion? Describe any four methods for its prevention.
- **19-** Describe discharge tube experiment with labeled diagram and What were the justification and conclusions of J.J. Thompson.

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